

ANNUAL REPORT
COMPREHENSIVE RESEARCH ON RICE
January 1, 2019 – December 31, 2019

PROJECT TITLE: The Environmental Fate of Pesticides Important to Rice Culture

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OBJECTIVES AND EXPERIMENTS CONDUCTED BY LOCATION TO ACCOMPLISH OBJECTIVES:

Objective I. To investigate the factors governing pesticide dissipation in California rice fields. Emphasis for 2019 will be to focus on the insecticide Coragen (chlorantraniliprole) by characterization of its degradation via sunlight (photolysis) under California rice field conditions.

Objective I. Photolysis of Coragen in Rice Field Soils

The photochemical degradation of chlorantraniliprole (3-Bromo-N-[4-chloro-2-methyl-6-(methylcarbamoyl)phenyl]-1-(3-chloro-2-pyridine-2-yl)-1H-pyrazole-5-carboxamide; CAP, Figure 1) was characterized under simulated solar light with 2-nitrobenzaldehyde (2NB) actinometry. Overall CAP degraded quickly via direct photolysis processes with no significant difference observed between high purity water and filtered rice field water. The 24 h average half life normalized to summer sunlight using 2NB actinometry was 34.45 ± 3.97 h ($j_{CAP,env} = 0.0203 \pm 0.0023$ h⁻¹, n = 3) and the calculated quantum yield was 0.00985 ± 0.000592 .

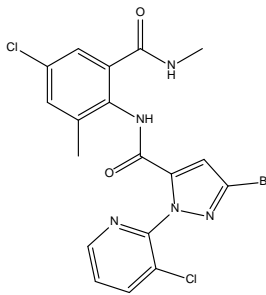


Figure 1. Chlorantraniliprole, the active ingredient of Coragen.

This work marks the completion of our goal to elucidate the environmental fate of chlorantraniliprole under California rice field conditions (Figure 2). Air-water and soil-water partitioning were previously characterized mathematically and via a batch equilibrium method. Air-water partitioning was characterized through calculation of Henry's law constant (K_H) between 15 – 35 °C while soil-water partitioning was investigated in three soils, collected from rice fields in the Sacramento Valley, at 15, 25 and 35 °C under varied aqueous salinities. K_H and organic carbon normalized soil water partitioning coefficient ($\text{Log}(K_{OC})$) determined for CAP were $1.69 \times 10^{-16} - 2.81 \times 10^{-15} \text{ atm} \cdot \text{m}^3 \cdot \text{mol}^{-1}$ and 2.59 – 2.93; indicating that volatilization will not contribute to its dissipation and its sorption to soils is moderately weak. Furthermore, sorption to CAP to soil was reversible resulting in an estimated 12 – 22% and 77 – 88 % of the applied mass to partition into water and soil, respectively, by the Pesticides in Flooded Applications Model (PFAM).

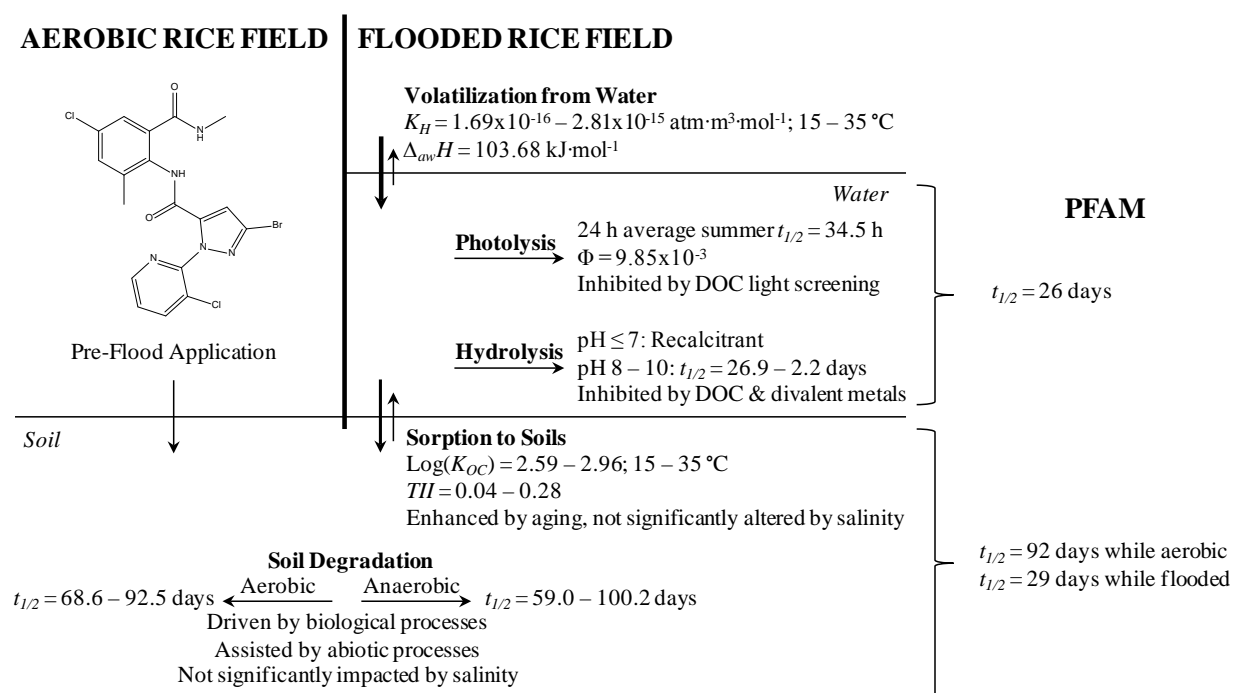


Figure 2. Summary of the environmental fate of CAP under simulated California rice field conditions.

As CAP would be applied pre-flooding directly to the soil surface and the majority of the applied mass is anticipated to remain in the soil its degradation in both unflooded and flooded rice field soils was investigated via microcosm experiments employing varied aqueous salt concentrations. The half-life ($t_{1/2}$) of CAP in two soils ranged from 59 – 100.2 days and was not significantly impacted by aqueous salinity; therefore, enhanced ionic strength due to evapoconcentration of salts in rice fields will not have a large impact on the dissipation of CAP from soils. However, degradation in sterilized microcosms ($t_{1/2} = 78.5 - 171.7 \text{ days}$) indicates that dissipation, while primarily biologically driven, was assisted by abiotic processes such as oxidation by manganese oxides and the formation of non-extractible residues. Overall degradation in soil was slow with 98% dissipation expected to occur within 471 – 601.2 days.

Due to its anticipated recalcitrance in soil, the hydrolysis of CAP was investigated within the pH range 6 – 10. Further experiments were conducted using buffered rice field water and the addition of divalent metals (Cu^{2+} , Zn^{2+} , Ni^{2+} , and Mn^{2+}) in order to determine the impact dissolved species common to California rice fields might have on the persistence of the insecticide. CAP was stable to hydrolysis under acidic and neutral conditions with addition of divalent metals having no observable catalytic effect; however, degradation occurred rapidly under alkaline conditions with half-lives ranging from 26.9 – 2.2 days between pH 8 – 10. Furthermore, the reaction was observed to proceed slower in buffered rice field water than lab grade water and the addition of Ni^{2+} and Mn^{2+} at elevated pH were observed to inhibit the reaction; indicating that the persistence of CAP, while dominated by pH, may be significantly altered between rice fields as a result of varied concentrations of dissolved organic carbon and metal species.

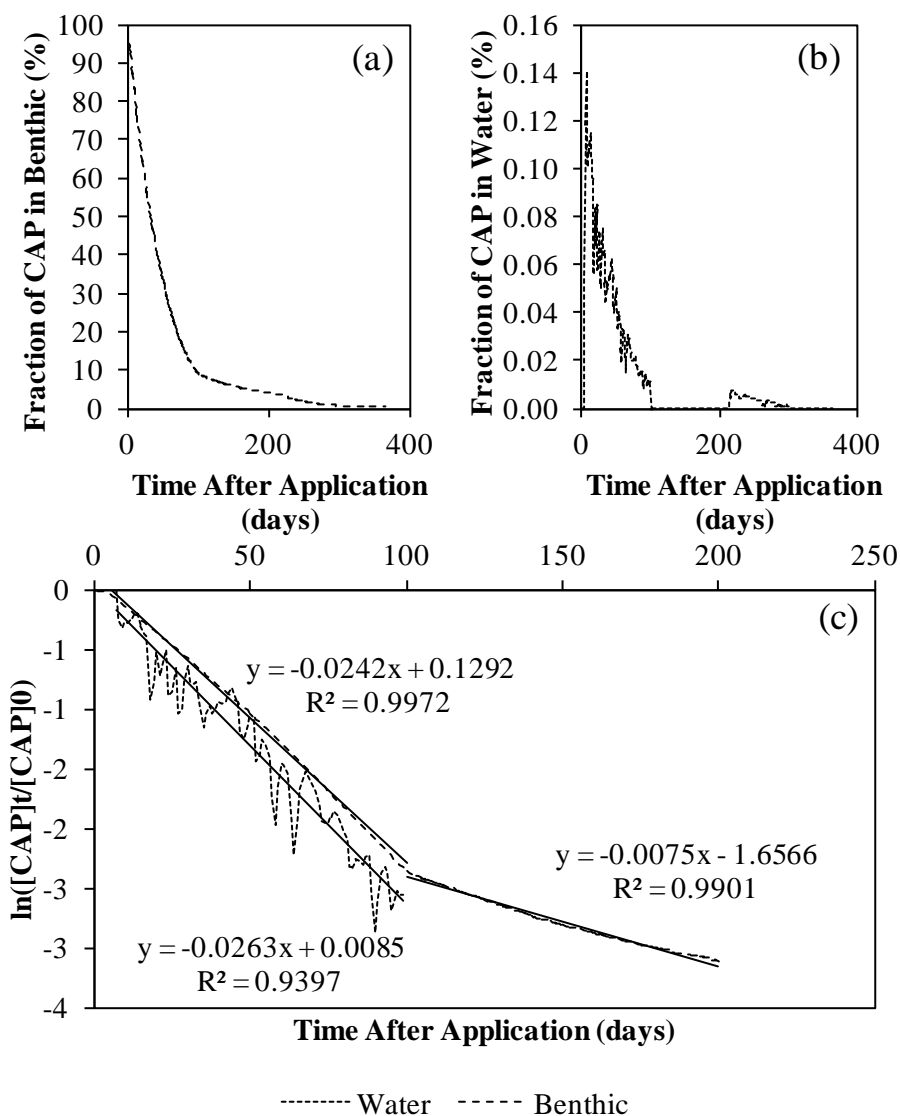


Figure 3. PFAM predicted dissipation of CAP from (a) benthic and (b) water compartments of a simulated California rice field with (c) natural-log linearized plot assuming first order degradation of CAP for the determination of estimated environmental half-lives.

Newly generated parameters for the photochemical degradation of CAP, reflective of California rice field conditions, were used alongside its previously characterized partitioning, degradation in soil and hydrolysis parameters in the Pesticides in Flooded Applications Model (PFAM) to simulate its dissipation in a model California rice field (Figure 3). The resulting simulation predicted that the majority of the insecticide will remain in the soil where it will degrade slowly ($t_{1/2} = 29$ and 92 days under flooded and non-flooded conditions) while the percentage of initially applied CAP in the water will remain below 0.2% as a result of its combined rapid photolysis and hydrolysis ($t_{1/2} = 26$ days). However, while the concentration of CAP in the aqueous phase is expected to remain low ($< 1 \mu\text{g}\cdot\text{L}^{-1}$), approximately 9% is expected to remain in the soil upon draining the field after a 100 day flooding period at which point its dissipation is expected to slow precipitously. As a result, CAP may accumulate over time in the soil leading to unintended chronic exposure to wildlife upon desorption during overwinter flooding or development of pest resistance particularly when considering that 67% remains in the soil after a more typical hold time of only 20 d.

Sample analysis via liquid chromatography-tandem mass spectrometry (LC-MS/MS) will continue into 2020 to confirm identification of the major degradation products resulting from both microbial and photochemical processes.

SUMMARY OF 2019 RESEARCH (major accomplishments) BY OBJECTIVE:

Objective I. The photolysis of chlorantraniliprole was investigated under California rice field conditions. CAP was degraded quickly via direct photolysis ($t_{1/2} = 34.5$ h) processes with no significant catalysis or inhibition by secondary processes in rice field water. The quantum yield was also determined using 2-nitrobenzaldehyde actinometry as 0.00985. These parameters were used along with partitioning, soil degradation and hydrolysis parameters previously determined in the Pesticides in Flooded Applications Model (PFAM). Simulation of the insecticide's overall fate in a California rice field resulted in an estimated environmental half-life of 26 days in the aqueous phase due to its rapid photolysis and hydrolysis; however, the majority of applied CAP is anticipated to persist in soils.

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CONCISE GENERAL SUMMARY OF CURRENT YEAR'S RESULTS:

1. The overall goal of our ongoing research program is to characterize the dissipation of pesticides under California rice field conditions. There are generally four contributing processes are investigated: volatilization to air, sorption (bonding) to soils, and degradation by either sunlight or soil microbes.
2. For Coragen, experiments conducted using simulated sunlight show that the active ingredient chlorantraniliprole (CAP) is expected to rapidly degrade via direct photolysis. Degradation in experiments using rice field water was not significantly different than those with high purity water, indicating that secondary photolysis will not be a major contributor to its dissipation. Chemical actinometry allowed for the derivation of environmentally relevant photochemical half-lives reflective of California rice field summer time growing conditions such that the Pesticides in Flooded Applications Model could be used to estimate its overall fate. Simulations estimate that CAP will degrade moderately fast once desorbed; however it is anticipated to persist in the soil for much longer periods.